## Organic compounds



1. Identify compound A. Show your working and justify your answers. [7]

Composition:
C 52.17\%; H 13.04\%; O 34.78\%
Mass spectra


Infra red spectrum


Use the information above to identify compound A:
2. Identify compound B. Show your working and justify your answers. [8]

Composition:
C $54.55 \%$ H $9.09 \%$ O 36.36\%
Mass spectra


## Infra red spectrum



Use the information above to identify compound B:
3. Identify compound C. Show your working and justify your answers. [8]

Composition:
C 40.00\%; H 6.67\%; O 53.33\%
Mass spectra


## Infra red spectrum



Use the information above to identify compound C:

Questions:
4. This question is about reaction 1 :
a. Write a balanced chemical reaction. [1]
b. 2.30 g of A reacted with an excess of the oxidising mixture to produce 2.00 g of B . Calculate the \% yield for this reaction. [5]
c. Calculate the atom economy. [1]
d. What is the oxidising mixture and state any colour changes you would see. [3]
e. How would you make B? Explain how this differs from reaction 2. [2]
5. This question is about reaction 3 :
a. Write a balanced chemical reaction. [1]
b. 2.30 g of A reacted with 3.50 g of C .4 .00 g of D was made. Calculate the $\%$ yield for this reaction. [6]
c. Calculate the atom economy. [1]
6. This question is about compound $E$
a. Identify compound E. Show your working and justify your answers.

## Composition:

C 37.21\%; H 7.75\%; CI 55.04\%

## Mass spectra


b. Explain the relatively large molecular ion peaks at 64 and 66? [1]
c. Use your knowledge from unit 1 to explain the actual Mr of E [1]
7. Use your knowledge of organic chemistry to identify F and G. [2]
8. Use your knowledge of organic chemistry to identify the types of reactions in reactions 4,5 and 6. [3]
9. Which of the reactions 4,5 and 6 will have the highest atom economy? Explain your answer. [2]
10. Pick one of the reactions, $1-6$ to draw a mechanism. [5]

