

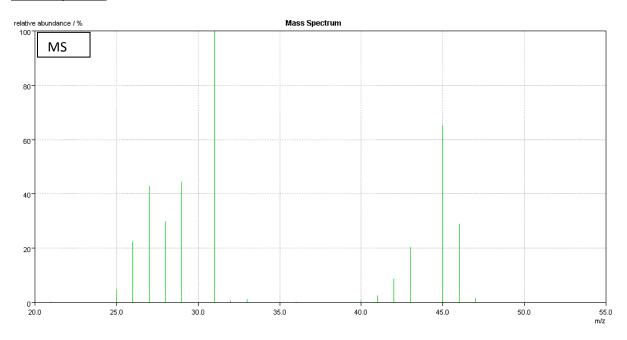
Answer the following questions: /58

1. Identify compound A. Show your working and justify your answers. [7]

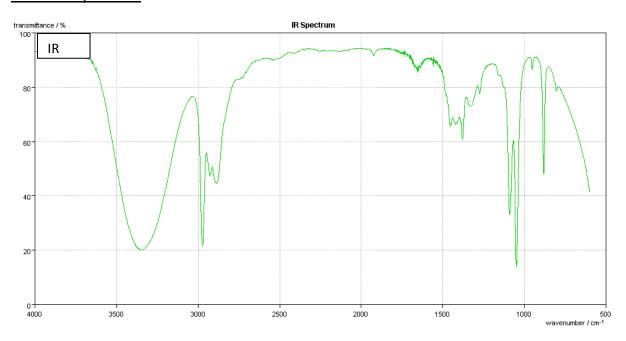
Composition:

C 52.17%; H 13.04%; O 34.78%

Mass spectra



Infra red spectrum



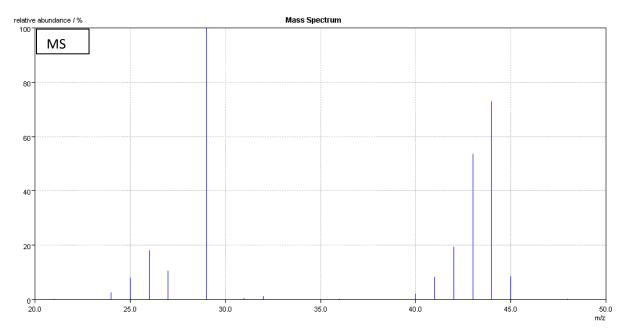
Use the information above to identify compound A:

2. Identify compound B. Show your working and justify your answers. [8]

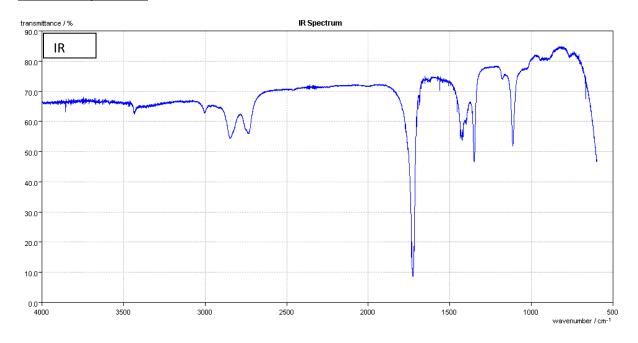
Composition:

C 54.55% H 9.09% O 36.36%

Mass spectra



Infra red spectrum



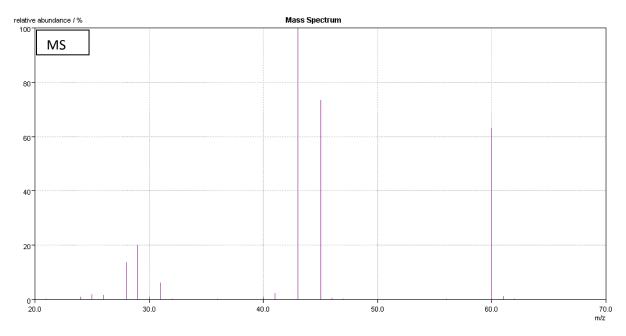
Use the information above to identify compound B:

3. Identify compound C. Show your working and justify your answers. [8]

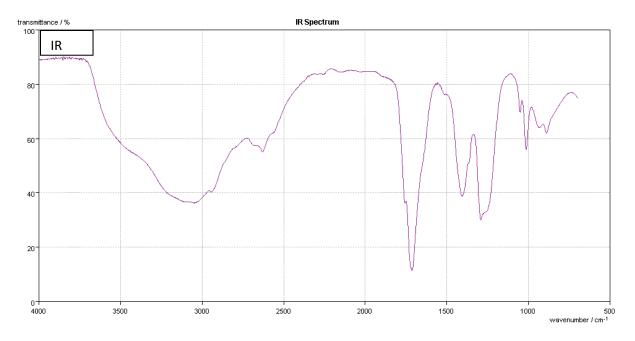
Composition:

C 40.00%; H 6.67%; O 53.33%

Mass spectra



Infra red spectrum



Use the information above to identify compound C:

Questions:

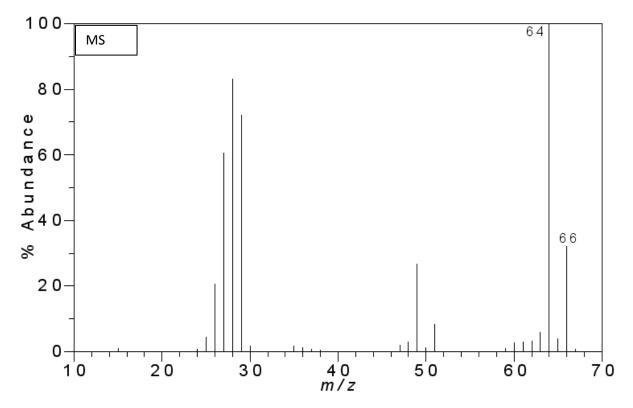
	uestion is about reaction 1: Write a balanced chemical reaction. [1]
b.	2.30g of A reacted with an excess of the oxidising mixture to produce 2.00g of B. Calculate the % yield for this reaction. [5]
C.	Calculate the atom economy. [1]
d.	What is the oxidising mixture and state any colour changes you would see. [3]
e.	How would you make B? Explain how this differs from reaction 2. [2]
•	uestion is about reaction 3: Write a balanced chemical reaction. [1]
b.	2.30g of A reacted with 3.50g of C. 4.00g of D was made. Calculate the % yield for this reaction. [6]
C.	Calculate the atom economy. [1]

- 6. This question is about compound E
 - a. Identify compound E. Show your working and justify your answers. [1]

Composition:

C 37.21%; H 7.75%; CI 55.04%

Mass spectra



- b. Explain the relatively large molecular ion peaks at 64 and 66? [1]
- c. Use your knowledge from unit 1 to explain the actual Mr of E [1]
- 7. Use your knowledge of organic chemistry to identify F and G. [2]
- 8. Use your knowledge of organic chemistry to identify the types of reactions in reactions 4,5 and 6. [3]
- 9. Which of the reactions 4,5 and 6 will have the highest atom economy? Explain your answer. [2]
- 10. Pick one of the reactions, 1 6 to draw a mechanism. [5]