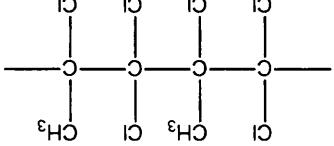


QUESTION PAPER

[Total: 13]

- (a) (i) $24.7/12 : 2.1/1 : 73.2/35.5 = 2.06 : 2.1 : 2.06$
- CHCl
- $48.5 \times 3 = 145.5$
- (ii) $(\text{CHCl} = 12 + 1 + 35.5 =) 48.5$
- Any two from
1. 1,1,3-trichloro
2. 1,3,3-trichloro
3. 3,3,3-trichloro
4. 2,3,3-trichloro
5. 1,3,3-trichloro
6. propene/prop-1-ene
7. correct numbers 1,2,3
8. marking points:
- (iii) 1,2,3-trichloropropene scores 1 mark ✓
 1 mark if backbone contains 4 carbons with 'end-bonds' and a reasonable attempt has been made
 e.g. used the wrong isomer.... max = 1 mark
- (c) (i) 
- any two gets 1 mark
- (iii) non-biodegradable
 toxic fumes evolved when burnt
 HCl or Cl₂ or chlorinated organic compounds such as COCl₂ also evolved when burnt

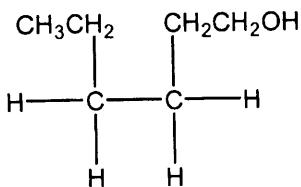
2812

Mark Scheme

January 2005

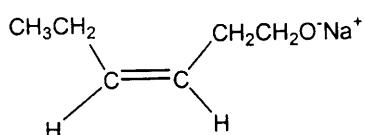
2.

(a) (i)



✓

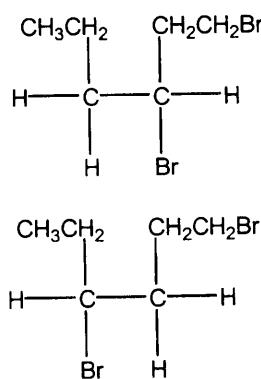
(ii)



charges are not necessary
allow the alkoxide ion

✓

(iii)

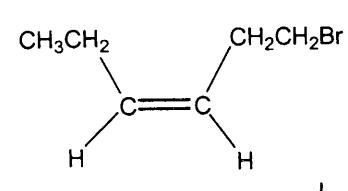
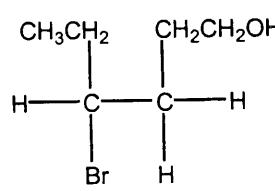
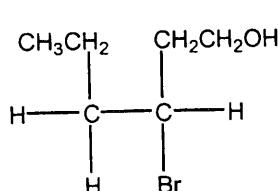


Both correct products gets 3 marks

✓✓✓

One correct product gets 2 marks

If neither of the above is correct then one mark can be awarded for any of:



max of two marks

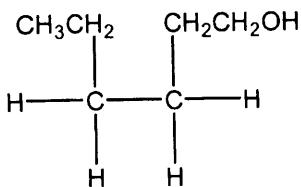
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Mark Scheme

January 2005

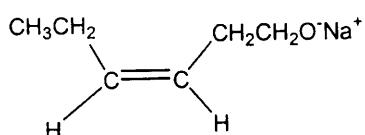
2.

(a) (i)



✓

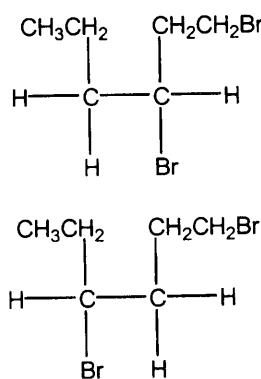
(ii)



charges are not necessary
allow the alkoxide ion

✓

(iii)

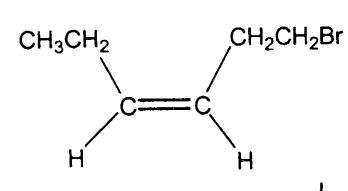
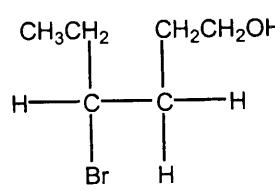
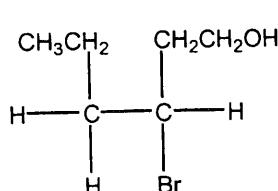


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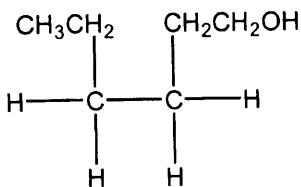
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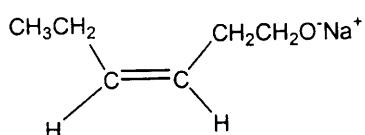
2.

(a) (i)



✓

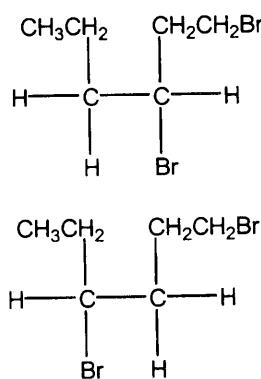
(ii)



charges are not necessary
allow the alkoxide ion

✓

(iii)

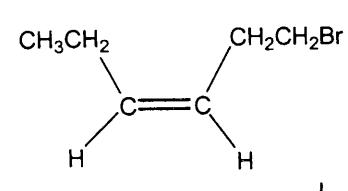
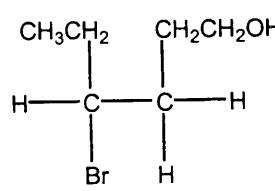
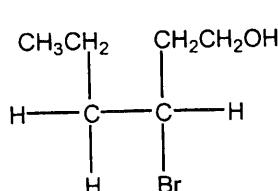


Both correct products gets 3 marks

✓✓✓

One correct product gets 2 marks

If neither of the above is correct then one mark can be awarded for any of:

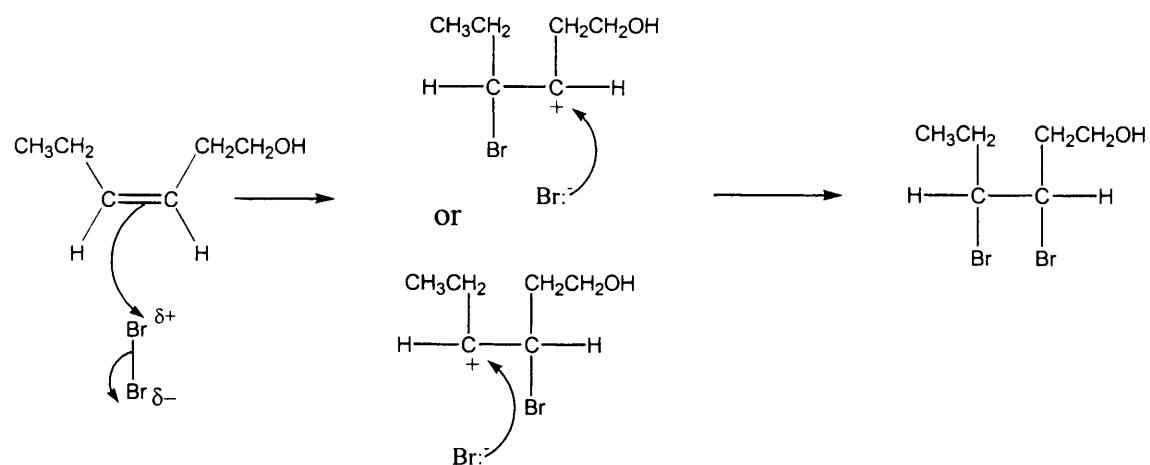


max of two marks

(b) (i) decolourises

✓

(ii)



curly arrow from C=C bond to bromine

✓

dipoles on Br₂ or curly arrow to show movement of bonded pair of electrons

✓

intermediate carbonium ion/carbocation

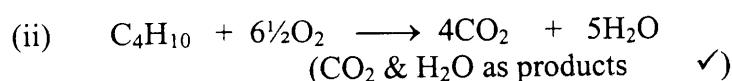
✓

curly arrow from lone pair on the Br⁻ ion to carbonium ion (Br^{δ-} loses 1 mark)

✓

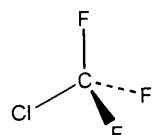
[Total: 10]

3. (a) (i) C_4H_{10}



(iii) propan-2-ol

(b) (i)



require an attempt at a 3D structure and
 bond angles must clearly not be 90°.

require at least one 'wedge' bond or one
 'dotted' bond

(ii) $108 - 111^\circ$

(iii) volatile/low boiling/gas/non-toxic/non-flammable/unreactive/liquefied under pressure/inert

(iv) homolytic = bonded pair split equally/ each retains 1 electron

fission = bond breaking

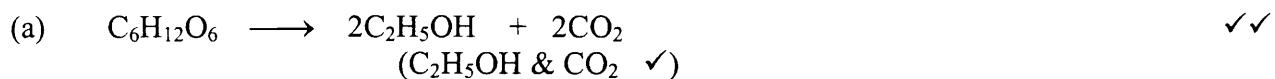
(v) C-Cl (no mark) because it is the weaker bond

(vi) Cl•

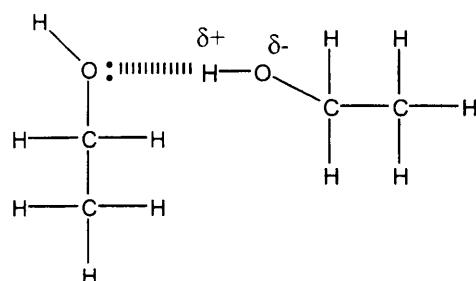
•CF₃ (allow CF₃•) (lack of 'dots' penalise once)

[Total: 12]

4.



(b)

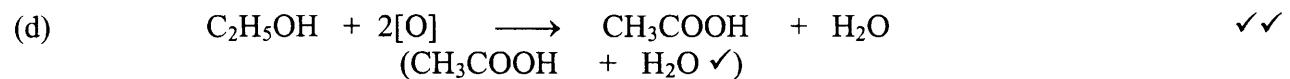


dipoles

hydrogen bond between O in one O-H and H in the other O-H ✓

lone pair from O involved in the H-bond ✓

- (c) (i) (volatile components) can escape/distil out ✓
 ethanal is most volatile/b pt less than 60 °C/partial oxidation ✓
- (ii) (volatile components) cannot escape/ refluxed ✓
 complete oxidation will be achieved/oxidised to the acid ✓



- (e) spectrum C ✓
 spectrum C only shows absorption at 1700 cm⁻¹ for the C=O ✓
 the other two spectra contain the OH group absorption at approx 3000 cm⁻¹ ✓

[Total: 14]

5.

identifies the three process as cracking, reforming, isomerisation ✓

recognises the need for high temperature or a catalyst ✓

equation for cracking ✓

equation for isomerisation ✓

state that reforming converts chains into rings/cyclic compounds ✓

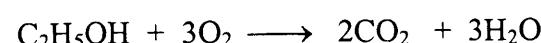
equation for reforming (balanced with H₂ could score two marks) ✓

sub-section mark = 6

oil is finite/non-renewable ✓

ethanol is renewable/sustainable ✓

from plants/crops/sugar cane/sugar beet/glucose/sugar/fermentation ✓



sub-section mark = 4

QWC

- organise relevant information clearly and coherently, using specialist vocabulary when appropriate (minimum of 4 from cracking/ isomerisation/ reforming/ renewable/ feedstock/ finite/fermentation/non-renewable/sustainable/zeolite/bimetallic catalyst/ etc) ✓
- reasonable spelling, punctuation and grammar throughout ✓

[Total: 11]